



education

Department of
Education
FREE STATE PROVINCE

CONTROL TEST / KONTROLETOETS

GRADE 11 / GRAAD 11

**PHYSICAL SCIENCES
FISIESE WETENSKAPPE**

MEMORANDUM

SEPTEMBER 2017

MARKS: 100 / PUNTE: 100

TIME: 2 HOURS / TYD: 2 URE

This memorandum consists of FIVE pages.
Hierdie memorandum bestaan uit VYF bladsye.

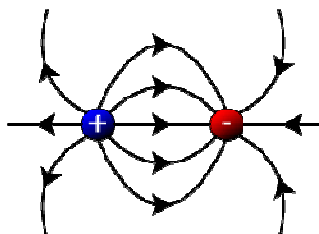
QUESTION 1/VRAAG 1

- 1.1 A ✓✓ (2)
1.2 C ✓✓ (2)
1.3 A ✓✓ (2)
1.4 D ✓✓ (2)
1.5 D ✓✓ (2)
1.6 A ✓✓ (2)
1.7 B ✓✓ (2)
1.8 A ✓✓ (2)
1.9 C ✓✓ (2)
1.1.0 D ✓✓ (2)

[20]

QUESTION 2/VRAAG 2

2.1.1



CRITERIA/KRITERIA	MARK PUNT
Shape of electric field; field lines do not touch or cross. <i>Patroon van elektriese veld; elektriese veldlyne raak of kruis nie.</i>	1
Direction of electric field (from positive to negative charge). <i>Rigting van elektriese veld (van positiewe na negatiewe lading).</i>	1

(2)

2.1.2 $E = \frac{kQ}{r^2} \checkmark = \frac{9 \times 10^9 \times 12 \times 10^{-6}}{0,2^2} \checkmark = 2,7 \times 10^6 \text{ N} \cdot \text{C}^{-1} \checkmark$

(3)

2.2.1 The magnitude of the electrostatic force exerted by one point charge (Q_1) on another point charge (Q_2) is directly proportional to the product of the magnitudes of the charges ✓ and inversely proportional to the square of the distance (r) between them. ✓ /

Die grootte van die elektrostatiese krag wat een puntlading (Q_1) op 'n ander puntlading uitoefen (Q_2) is direk eweredig aan die produk van die groottes van die ladings ✓ en omgekeerd eweredig aan die kwadraat van die afstand (r) tussen hulle. ✓

(2)

2.2.2 $F = \frac{kQ_1Q_2}{r^2} \checkmark$

$F = \frac{(9 \times 10^9)(14 \times 10^{-6})(10 \times 10^{-6})}{(0,1)^2} \checkmark = 126 \text{ N}$

$F = \frac{kQ_2Q_3}{r^2}$

$F = \frac{(9 \times 10^9)(12 \times 10^{-6})(10 \times 10^{-6})}{(0,15)^2} \checkmark = 48 \text{ N}$

$F_{\text{net}} = \sqrt{126^2 + 48^2} \checkmark = 134,83 \text{ N} \checkmark$

(6)

[13]

QUESTION 3/VRAAG 3

3.1.1 Galvano meter / Galvanometer ✓ (1)

3.1.2 Y to X / Y na X (✓✓) (2)

3.1.3 S pole / S-pool ✓ (1)

3.2 The magnitude of the induced emf induced across the ends of a conductor is directly proportional ✓ to the rate of change in the magnetic flux linkage with the conductor. ✓ /

Die grootte van die geïnduseerde emk oor die ente van 'n geleier is direk eweredig ✓ aan die tempo van verandering van magnetiese vloedkoppeling met die geleier. ✓ (2)

3.3 $\Delta\Phi = (B_f - B_i)A \cos \theta \checkmark = (1 - 0,5)(0,5)^2 \cos 0^\circ \checkmark = 0,125 \text{ Wb}$

$$\varepsilon = -N \frac{\Delta\Phi}{\Delta t} \checkmark = -5 \times \frac{0,125}{10} \checkmark = -0,06 \text{ V} \checkmark$$

Accept if negative is not shown. / Aanvaar as negatief nie gewys is nie. (6)

[12]

QUESTION 4/VRAAG 4

4.1 Current / Stroom ✓ (1)

4.2.1 (4;0,63) Allow for y between / Aanvaar vir y tussen: 0,62 - 0,64 ✓ ✓ (2)

4.2.2 Temperature increases. / Temperatuur verhoog. ✓
R increase; gradient decrease / R neem toe; gradiënt neem af. ✓ (2)

4.3 Gradient = $\frac{\Delta I}{\Delta V} = \frac{0,63 - 0}{4 - 0} \checkmark = 0,158 \text{ A} \cdot \text{V}^{-1} (\Omega^{-1})$

$$R = \frac{1}{0,158} \checkmark = 6,35 \Omega \checkmark$$

(4)

[9]

QUESTION 5/VRAAG 5

5.1 $R = \frac{V}{I} \checkmark = \frac{20}{4} \checkmark = 5 \Omega \checkmark$ (3)

5.2 $V_2 = IR = 4 \checkmark \times 3,2 \checkmark = 12,8 \text{ V}$ $V_1 = 20 - 12,8 \checkmark = 7,2 \text{ V} \checkmark$ (4)

5.3

OPTION 1 / OPSIE 1 Positive marking from 5.2. <i>Positiewe nasien vanaf 5.2.</i>	OPTION 2 / OPSIE 2 Positive marking from 5.1. <i>Positiewe nasien vanaf 5.1.</i>
$I_{3\Omega} = \frac{V}{R} \checkmark = \frac{7,2}{3} \checkmark = 2,4 \text{ A}$ $I_R = I_T - I_{3\Omega} = 4 - 2,4 \checkmark = 1,6 \text{ A}$ $R = \frac{V}{I} = \frac{7,2}{1,6} \checkmark = 4,5 \Omega \checkmark$	$R_P = R_{TOT} - R_{3,2\Omega}$ $= 5 - 3,2 \checkmark = 1,8 \Omega$ $\frac{1}{R_{//}} = \frac{1}{R} + \frac{1}{R_3} \checkmark \Rightarrow \checkmark \frac{1}{1,8} = \frac{1}{R} + \frac{1}{3} \checkmark$ $\frac{1}{R} = \frac{2}{9} \quad R = 4,5 \Omega \checkmark$

(5)

5.4 **Positive marking from 5.1 to 5.3/Positiewe nasien vanaf 5.1 tot 5.3.**

OPTION 1 / OPSIE 1:	OPTION 2 / OPSIE 2:	OPTION 3 / OPSIE 3:
$P = I^2 R \checkmark$ $= (2,4^2)(3) \checkmark$ $= 17,28 \text{ W} \checkmark$	$P = \frac{V^2}{R} \checkmark = \frac{7,2^2}{3} \checkmark = 17,28 \text{ W} \checkmark$	$P = VI \checkmark$ $= 7,2 \times 2,4 \checkmark$ $= 17,28 \text{ W} \checkmark$

(3)

5.5 $\text{Cost/Koste} = \frac{17,28}{1000} \checkmark \times 10 \times R2,20 \checkmark = R0,38 \checkmark$ (3)

[18]**QUESTION 6/VRAAG 6**

6.1 The empirical formula is the simplest whole number ratio $\checkmark$ of atoms in a compound. $\checkmark$
 Die empiriese formule is die eenvoudigste heelgetalverhouding $\checkmark$ van atome in 'n verbinding. $\checkmark$ (2)

6.2

	Na	:	Cr	:	O	
	17,5	:	39,7	:	42,8	
n(mol)	$\frac{17,5}{23}$	:	$\frac{39,7}{52}$	:	$\frac{42,8}{16}$	$\checkmark$
n(mol)	0,76	:	0,76	:	2,675	$\checkmark$
	1	:	1	:	3,5	$\checkmark$
	2	:	2	:	7	$\checkmark$

Formula:/Formule: $\text{Na}_2\text{Cr}_2\text{O}_7 \checkmark$

(5)

[7]

QUESTION 7/VRAAG 7

$$7.1 \quad n(\text{MgCO}_3) = \frac{m}{M} \checkmark = \frac{126}{84} \checkmark = 1,5 \text{ mol} \checkmark \quad (3)$$

$$7.2 \quad m(\text{MgCl}_2) = n \times M = 3 \checkmark \times 95 \checkmark = 285 \text{ g} \checkmark \quad (3)$$

$$7.3 \quad V_{(\text{CO}_2)} = n \times V_m \checkmark = 6 \times 22,4 \checkmark = 134,4 \text{ dm}^3 \checkmark \quad (3)$$

$$7.4 \quad V_{(\text{HCl})} = \frac{n}{c} \checkmark = \frac{0,4 \checkmark}{0,5 \checkmark} = 0,8 \text{ dm}^3 \checkmark \quad (4)$$

[13]

QUESTION 8/VRAAG 8

8.1

$$n(\text{TiCl}_4) = \frac{m}{M} = \frac{3\,540\,000}{190} \checkmark = 18\,631,58 \text{ mol}$$

$$n(\text{Mg}) = \frac{m}{M} = \frac{1\,130\,000}{24} \checkmark = 47\,083,33 \text{ mol}$$

$$n(\text{Mg}) = 2 \times n(\text{TiCl}_2) \checkmark$$

$$= 2 \times 18\,631,58 \checkmark$$

$$= 37\,263,16 \text{ mol} \checkmark$$

$\therefore$ Mg in excess/Mg in oormaat $\checkmark$

OR/OF

$$n(\text{TiCl}_4) = 0,5 \times n(\text{Mg}) \checkmark$$

$$= 0,5 \times 47\,083,33 \checkmark$$

$$= 23\,541,67 \text{ mol} \checkmark$$

$\therefore$ TiCl₄ shortage/tekort $\checkmark$

$$n(\text{Ti}) = \frac{m}{M} = \frac{894,32 \times 1000 \checkmark}{48 \checkmark}$$

$$= 18631,667 \text{ mol}$$

$$n(\text{TiCl}_4) = \frac{m}{M}$$

$$18631,667 \checkmark = \frac{m}{190 \checkmark}$$

$$m = 3540,017 \text{ kg} \checkmark$$

TiCl₄ is used up/word opgebruik. $\checkmark$

OR/OF

$$n(\text{Mg}) = \frac{m}{M}$$

$$2 \times 18631,667 \checkmark = \frac{m}{24 \checkmark}$$

$$m = 894,32 \text{ kg} \checkmark$$

Mg in excess/Mg in oormaat $\checkmark$

(6)

8.2

$$\% \text{ Yield Ti} = \frac{\text{Real yield}}{\text{Theoretical yields}}$$

$$= \frac{820}{894,32} \checkmark = 91,69\% \checkmark$$

$$\% \text{ Opbrengs Ti} = \frac{\text{Werklike opbrengs}}{\text{Teoretiese opbrengs}}$$

$$= \frac{820}{894,32} \checkmark = 91,69\% \checkmark$$

(2)

[8]

GRAND TOTAL/GROOTTOTAAL : 100